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QC100 .U573 V34:1970 C.1 NBS-PUB-C 1964

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Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra

Charlotte E. Moore

Office of Standard Reference Data
National Bureau of Standards
Washington, D.C. 20234



t, **NSRDS-NBS 34**

Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 34, 22 pages (Sept. 1970)

CODEN: NSRDA

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Issued September 1970

For sale by the Superintendent of Documents, U.S. Government Printing Office
Washington, D.C., 20402 (Order by SD Catalog No. C 13.48:34) - Price 75 cents

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FOREWORD

The National Standard Reference Data System provides effective access to the quantitative data of physical science, critically evaluated and compiled for convenience, and readily accessible through a variety of distribution channels. The System was established in 1963 by action of the President's Office of Science and Technology and the Federal Council for Science and Technology, with responsibility to administer it assigned to the National Bureau of Standards.

The System now comprises a complex of data centers and other activities, carried on in academic institutions and other laboratories both in and out of government. The independent operational status of existing critical data projects is maintained and encouraged. Data centers that are components of the NSRDS produce compilations of critically evaluated data, critical reviews of the state of quantitative knowledge in specialized areas, and computations of useful functions derived from standard reference data. In addition, the centers and projects establish criteria for evaluation and compilation of data and make recommendations on needed improvements in experimental techniques. They are normally closely associated with active research in the relevant field.

The technical scope of the NSRDS is indicated by the principal categories of data compilation projects now active or being planned: nuclear properties, atomic and molecular properties, solid state properties, thermodynamic and transport properties, chemical kinetics, and colloid and surface properties.

The NSRDS receives advice and planning assistance from the National Research Council of the National Academy of Sciences-National Academy of Engineering. An overall Review Committee considers the program as a whole and makes recommendations on policy, long-term planning, and international collaboration. Advisory Panels, each concerned with a single technical area, meet regularly to examine major portions of the program, assign relative priorities, and identify specific key problems in need of further attention. For selected specific topics, the Advisory Panels sponsor subpanels which make detailed studies of users' needs, the present state of knowledge, and existing data resources as a basis for recommending one or more data compilation activities. This assembly of advisory services contributes greatly to the guidance of NSRDS activities.

The NSRDS-NBS series of publications is intended primarily to include evaluated reference data and critical reviews of long-term interest to the scientific and technical community.

LEWIS M. BRANSCOMB, *Director*

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Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra

Charlotte E. Moore

A current table of ionization potentials expressed in electron volts and a detailed table giving the limits from which they have been derived are presented. For each spectrum the ground term is given, with the limit as the ground state. The energy levels of terms of the lowest configuration determined from ground state zero, are also included for selected spectra. The literature references used for each spectrum are indicated by number and listed in a bibliography with some 200 entries.

The latest recommended conversion factor (cm^{-1} to eV) 0.000123981 corresponding to $1 \text{ eV} = 8065.73 \text{ cm}^{-1}$ has been used throughout.

Key words: Atomic spectra, ground terms; ground terms, atomic spectra; ionization limits; ionization potentials.

The data in the Volumes on "Atomic Energy Levels" (AEL) [135], [136], [137], include the ionization limits known for individual spectra. The latest table of ionization potentials calculated from these limits was published as Table 34 in Volume III (1958). Much work has been done since then and there has been a steady demand for a revision of this Table.

A fairly comprehensive general bibliography has recently been published [194] which lists for each spectrum the literature references on analyses of atomic spectra dating from the entries in the respective Volume of "AEL" (1949), (1952), (1958), well into 1968. The present compendium is based largely on the references in this Bibliography, with some, but probably not all, later material.

The reliability of the data recorded in the literature is often difficult to appraise. In cases where long series are known in the various spectra, the ionization potentials are well determined. With these as key points, good values can be derived by extrapolation or interpolation along isoelectronic sequences, or by comparison along the rows in the Periodic Chart for spectra of similar stages of ionization. Frequently, however, authors give values of ionization potentials without stating the conversion factor used and without describing clearly how the quoted value was obtained.

For this reason, the present paper includes not only the ionization potentials in eV, but also, the limits in cm^{-1} from which these have been derived. Table 1 gives the ionization potentials in eV for each spectrum.

The conversion factor taken from [195] was used for Table 1, since it is the value currently recommended by the National Academy of Sciences-National Research Council. However, recent measurements [200] suggest that this value may be in error by about 30 parts per million. Therefore, it should be understood that all of the significant figures included in Table 1 may not be meaningful

in an absolute sense. This applies particularly to entries with magnitudes greater than 100 eV.

All limits have been multiplied by the factor 0.000123981 to obtain the entries in Table 1, i.e., $1 \text{ eV} = 8065.73 \text{ cm}^{-1}$. The factor used in "AEL" was 0.00012395 and has been superseded. As a result, in the present table there are systematic differences from the 1958 Table, caused by the change in the conversion factor, as well as the differences caused by improved values of the limits.

Italics denote ionization potentials derived from limits that are bracketed in Table 2.

In compiling Table 1 the author has attempted to indicate roughly the various degrees of accuracy of the limits. Those based on well-established series deserve the greatest weight. When the ionization potential is given to three places, it is felt that the third place is meaningful. The two- and one-place entries are less well defined, but it is hoped that they have some significance. The limits of error assigned by the various investigators provide a general criterion, but these are given for comparatively few spectra. Users should, therefore, consult the limits given in Table 2 and the references in order to evaluate the data for individual spectra.

Table 2 contains the basic data for each spectrum. As in Table 1, the successive stages of ionization are indicated at the heading of each column: I, denoting first spectra (neutral atoms); II, second spectra (singly ionized atoms), etc. The elements are arranged in order of increasing atomic number, Z . The ground state is indicated for each spectrum, together with the ionization limit in cm^{-1} . In every case this limit refers to the ground state of the ion in the next higher stage of ionization. The limits of error are quoted from the original authors. Although not specifically defined, these afford a general guide as to the reliability of the limit.

Although all limits are based on data derived from the analyses of optical spectra, they are determined in various ways, since reliable series are

TABLE I. Ionization potentials*

Z	Element	Spectrum									
		I	II	III	IV	V	VI	VII	VIII	IX	X
1	H	13.598									
2	He	24.587	54.416								
3	Li	5.392	75.638	122.451							
4	Be	9.322	18.211	153.893	217.713						
5	B	8.298	25.154	37.930	259.368	340.217					
6	C	11.260	24.383	47.887	64.492	392.077	489.981				
7	N	14.534	29.601	47.448	77.472	97.888	552.057	667.029			
8	O	13.618	35.116	54.934	77.412	113.896	138.116	739.315	871.387		
9	F	17.422	34.970	62.707	87.138	114.240	157.161	185.182	953.886	1103.089	
10	Ne	21.564	40.962	63.45	97.11	126.21	157.93	207.27	239.09	1195.797	1362.164
11	Na	5.139	47.286	71.64	98.91	138.39	172.15	208.47	264.18	299.87	1465.091
12	Mg	7.646	15.035	80.143	109.24	141.26	186.50	224.94	265.90	327.95	367.53
13	Al	5.986	18.828	28.447	119.99	153.71	190.47	241.43	284.59	330.21	398.57
14	Si	8.151	16.345	33.492	45.141	166.77	205.05	246.52	303.17	351.10	401.43
15	P	10.486	19.725	30.18	51.37	65.023	220.43	263.22	309.41	371.73	424.50
16	S	10.360	23.33	34.83	47.30	72.68	88.049	280.93	328.23	379.10	447.09
17	Cl	12.967	23.81	39.61	53.46	67.8	97.03	114.193	348.28	400.05	455.62
18	Ar	15.759	27.629	40.74	59.81	75.02	91.007	124.319	143.456	422.44	478.68
19	K	4.341	31.625	45.72	60.91	82.66	100.0	117.56	154.86	175.814	503.44
20	Ca	6.113	11.871	50.908	67.10	84.41	108.78	127.7	147.24	188.54	211.270
21	Sc	6.54	12.80	24.76	73.47	91.66	111.1	138.0	158.7	180.02	225.32
22	Ti	6.82	13.58	27.491	43.266	99.22	119.36	140.8	168.5	193.2	215.91
23	V	6.74	14.65	29.310	46.707	65.23	128.12	150.17	173.7	205.8	230.5
24	Cr	6.766	16.50	30.96	49.1	69.3	90.56	161.1	184.7	209.3	244.4
25	Mn	7.435	15.640	33.667	51.2	72.4	95	119.27	196.46	221.8	243.3
26	Fe	7.870	16.18	30.651	54.8	75.0	99	125	151.06	235.04	262.1
27	Co	7.86	17.06	33.50	51.3	79.5	102	129	157	186.13	276
28	Ni	7.635	18.168	35.17	54.9	75.5	108	133	162	193	224.5
29	Cu	7.726	20.292	36.83	55.2	79.9	103	139	166	199	232
30	Zn	9.394	17.964	39.722	59.4	82.6	108	134	174	203	238
31	Ga	5.999	20.51	30.71	64						
32	Ge	7.899	15.934	34.22	45.71	93.5					
33	As	9.81	18.633	28.351	50.13	62.63	127.6				
34	Se	9.752	21.19	30.820	42.944	68.3	81.70	155.4			
35	Br	11.814	21.8	36	47.3	59.7	88.6	103.0	192.8		
36	Kr	13.999	24.359	36.95	52.5	64.7	78.5	111.0	126	230.9	
37	Rb	4.177	27.28	40	52.6	71.0	84.4	99.2	136	150	277.1
38	Sr	5.695	11.030	43.6	57	71.6	90.8	106	122.3	162	177
39	Y	6.38	12.24	20.52	61.8	77.0	93.0	116	129	146.2	191
40	Zr	6.84	13.13	22.99	34.34	81.5					
41	Nb	6.88	14.32	25.04	38.3	50.55	102.6	125			
42	Mo	7.099	16.15	27.16	46.4	61.2	68	126.8	153		

TABLE I. *Ionization potentials** – Continued

Spectrum – Continued											Z
XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	XX	XXI	
											1
											2
											3
											4
											5
											6
											7
											8
											9
											10
1648.659											11
1761.802	1962.613										12
442.07	2085.983	2304.080									13
476.06	523.50	2437.676	2673.108								14
479.57	560.41	611.85	2816.943	3069.762							15
504.78	564.65	651.63	707.14	3223.836	3494.099						16
529.26	591.97	656.69	749.74	809.39	3658.425	3946.193					17
538.95	618.24	686.09	755.73	854.75	918	4120.778	4426.114				18
564.13	629.09	714.02	787.13	861.77	968	1034	4610.955	4933.931			19
591.25	656.39	726.03	816.61	895.12	974	1087	1157	5129.045	5469.738		20
249.832	685.89	755.47	829.79	926.00							21
265.23	291.497	787.33	861.33	940.36							22
255.04	308.25	336.267	895.58	974.02							23
270.8	298.0	355	384.30	1010.64							24
286.0	314.4	343.6	404	435.3	1136.2						25
290.4	330.8	361.0	392.2	457	489.5	1266.1					26
305	336	379	411	444	512	546.8	1403.0				27
321.2	352	384	430	464	499	571	607.2	1547			28
266	368.8	401	435	484	520	557	633	671	1698		29
274	310.8	419.7	454	490	542	579	619	698	738	1856	30
											31
											32
											33
											34
											35
											36
											37
324.1											38
206	374.0										39
											40
											41
											42

TABLE I. *Ionization potentials** – Continued

Z	Element	Spectrum									
		I	II	III	IV	V	VI	VII	VIII	IX	X
43	Tc	7.28	15.26	29.54							
44	Ru	7.37	16.76	28.47							
45	Rh	7.46	18.08	31.06							
46	Pd	8.34	19.43	32.93							
47	Ag	7.576	21.49	34.83							
48	Cd	8.993	16.908	37.48							
49	In	5.786	18.869	28.03	54						
50	Sn	7.344	14.632	30.502	40.734	72.28					
51	Sb	8.641	16.53	25.3	44.2	56	108				
52	Te	9.009	18.6	27.96	37.41	58.75	70.7	137			
53	I	10.451	19.131	33							
54	Xe	12.130	21.21	32.1							
55	Cs	3.894	25.1								
56	Ba	5.212	10.004								
57	La	5.577	11.06	19.175							
58	Ce	5.47	10.85	20.20	36.72						
59	Pr	5.42	10.55	21.62	38.95	57.45					
60	Nd	5.49	10.72								
61	Pm	5.55	10.90								
62	Sm	5.63	11.07								
63	Eu	5.67	11.25								
64	Gd	6.14	12.1								
65	Tb	5.85	11.52								
66	Dy	5.93	11.67								
67	Ho	6.02	11.80								
68	Er	6.10	11.93								
69	Tm	6.18	12.05	23.71							
70	Yb	6.254	12.17	25.2							
71	Lu	5.426	13.9								
72	Hf	7.0	14.9	23.3	33.3						
73	Ta	7.89									
74	W	7.98									
75	Re	7.88									
76	Os	8.7									
77	Ir	9.1									
78	Pt	9.0	18.563								
79	Au	9.225	20.5								
80	Hg	10.437	18.756	34.2							
81	Tl	6.108	20.428	29.83							
82	Pb	7.416	15.032	31.937	42.32	68.8					
83	Bi	7.289	16.69	25.56	45.3	56.0	88.3				

TABLE I. Ionization potentials* — Continued

Z	Element	Spectrum					
		I	II	III	IV	V	
84	Po	8.42					
86	At						
86	Rn	10.748					
87	Fr						
88	Ra	5.279	10.147				
89	Ac	6.9	12.1				
90	Th		11.5	20.0	28.8		
91	Pa						
92	U						
93	Np						
94	Pu	5.8					
95	Am	6.0					

* $1\text{cm}^{-1} = 0.000123981\text{ eV}$.

known for only a limited number of spectra. For the H I and He I isoelectronic sequences, the theoretical values quoted here are well determined. Edlén, [44], [45], [46], [47], has made a detailed study of formulae for extrapolating ionization limits along sequences of the lighter elements. His values are extensively quoted in Table 2.

Catalán and his associates, [22 to 27], have interpolated values for spectra of neighboring elements in the same stage of ionization. These have been used for spectra in which series are not known. Russell, [166], Sugar and Reader, [156], [181] and others, have described similar general relationships between spectra, that can be used to derive fairly reliable limits.

In Table 2 all ionization limits were recorded that were derived from observed series, from extrapolation or interpolation as described above (Edlén, Catalán, etc.), or from theoretical calculations such as those of the H I and He I series. When all available data from these sources had been entered, if gaps still remained for spectra of a given element in successive stages of ionization, the intervening limits were entered in brackets, as for Ti VIII and Ti IX. These limits, in brackets, represent calculated values interpolated or extrapolated from observed data, and reported in two general tables of ionization potentials in which different methods have been used. For scattered spectra of the elements S V through Zn XIX, the table of Lotz, [116], has been quoted. For larger atomic numbers, the entries in brackets are from the table of Finkelnburg and Humbach, [65]. No attempt has been made, however, to quote *all* such calculated values.

The need for higher ionization limits within a given spectrum increases as laboratory research on absorption series in the vacuum ultraviolet, on series produced with synchrotron radiation as a

source, and the like, advances. At the request of workers in these fields, all components of the ground term, and in selected cases, all levels from the ground configuration, are entered in Table 2. All levels above the ground state are relative to the ground state zero. For example, in the format of "AEL," the lowest levels of O I are as follows:

Desig.	AEL	Table 2
$2p^4\ ^3P_2$	0.000	109837.02 = Limit
$\ ^3P_1$	158.265	158.265
$\ ^3P_0$	226.977	226.977
$\ ^1D_2$	15867.862	15867.862
$\ ^1S_0$	33792.583	33792.583

In compiling Table 2, the energy levels of *only* the ground term have been included for complex spectra, particularly with increasing Z. It is well known that in rare-earth spectra low configurations and low terms overlap in many cases. Consequently, many more low energy levels may be known than those of the ground term. Users are urged to recognize this limitation of the Table and to consult the literature references for further details concerning the low levels that have been reported for individual spectra.

As in "AEL" estimated values of energy levels are given in brackets. Similarly, "x" denotes that the energy level is not connected by observation with the others.

In Table 2, under the term designations for each spectrum, the numbers in italics at the lower left, refer to Table 3. This table is a Bibliography which contains the literature references used for each spectrum to obtain the limits and terms quoted in Table 2.

The importance of stating, clearly, how a limit or an ionization potential has been derived cannot be overemphasized. It is hoped that the present tables will enable each user to judge the quality of the available data used to compile Table 1.

Although the foregoing results are limited to optical spectra, it should be recognized that experimental values of ionization energies have, also, been published. A surface ionization method has been used to obtain ionization potentials for first spectra of rare earths, [196 to 198]. In general, the agreement is satisfactory between the values obtained by the different methods.

Estimates of ionization potentials of third spectra of the lanthanons have been calculated recently "by applying the Born-Haber cycle to the group 3A oxides and arsenides." [199].

After the work on the present publication had been started, the author learned that extensive revisions of the data on the spectra of lighter elements were being prepared by B. Edlén, J. O. Ekberg, and L. Å. Svensson, in Lund. They have most generously furnished much valuable material, in advance of publication, for inclusion here. The author is deeply indebted to these colleagues whose expert judgment and advice greatly enhance the value of the present publication. She is equally grateful to all others who have so willingly contributed their unpublished material.

Washington, D.C.
April 22, 1970

Table 3. Bibliography

References

1. Albertson, W., *Astroph. J.* **84**, No. 1, 26-72 (1936).
2. Albertson, W., *Phys. Rev.* **52**, 644-647 (1937).
3. Alexander, E., Feldman, U., and Fraenkel, B. S., *J. Opt. Soc. Am.* **55**, No. 6, 650-653 (1965).
4. Alexander, E., Feldman, U., Fraenkel, B. S., and Hoory, S., *J. Opt. Soc. Am.* **56**, No. 5, 651-652 (1966).
5. Andersson, E., and Johannesson, G. A., unpublished material (1969).
6. Andrew, K. L., and Meissner, K. W., *J. Opt. Soc. Am.* **49**, No. 2, 146-161 (1959).
7. Bauche, J., Blaise, J., and Fred, M., *J. Opt. Soc. Am.* **54**, No. 4, 565 (A) (1964).
8. Blaise, J., Fred, M., Gerstenkorn, S., and Judd, B. R., *Compt. Rend.* **255**, 2403-2405 (1962).
9. Bockasten, K., *Ark. Fys. (Stockholm)* **10**, No. 40, 567-582 (1956).
10. Bockasten, K., unpublished material (1963).
11. Bockasten, K., *J. Opt. Soc. Am.* **54**, No. 8, 1065 (L) (1964).
12. Bockasten, K., unpublished material (1969), based on unpublished data communicated by Wu, C. M., and Crooker, A. M.
13. Bockasten, K., and Edlén, B., private communication (1970).
14. Bockasten, K., Hallin, R., and Hughes, T. P., *Proc. Phys. Soc. (London)* **81**, Part 3, No. 521, 522-530 (1963).
15. Bockasten, K., and Johansson, K. B., *Ark. Fys. (Stockholm)* **38**, No. 31, 563-584 (1968).
16. Bórgström, A., *Ark. Fys. (Stockholm)* **38**, No. 12, 243-260 (1968).
17. Bowen, I. S., *Astroph. J.* **121**, No. 2, 306-311 (1955); **132**, No. 1, 1-17 (1960).
18. Bromander, J., unpublished material (Uppsala 1969), from data by Glad, S.; See *Ark. Fys. (Stockholm)* **7**, No. 2, 7 to 32 (1954).
19. Bromander, J., and Johansson, B., *J. Opt. Soc. Am.* **57**, No. 9, 1158-1159 (L) (1967).
20. Bryant, B. W., *J. Opt. Soc. Am.* **55**, No. 7, 771-779 (1965), with revision by J. Sugar (1970).
- 20a. Camus, P. et Tomkins, F. S., *J. de Phys.* **30**, No. 7, 545-550 (1969).
- 20b. Camus, P. et Tomkins, F. S., *Première Réunion Ann. de l'Assoc. Européenne de Spectroscopie Atomique*, Abstract 67, Paris-Orsay (1969).
21. Catalán, M. A., Meggers, W. F., and Garcia-Riquelme, O., *J. Res. Nat. Bur. Stand. (U.S.)*, **68A** (Phys. and Chem.) No. 1, 9-59 (Jan.-Feb. 1964).
22. Catalán, M. A. y Rico, F. R., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **48**, Nos. 11, 12, 328-338 (1952).
23. Catalán, M. A. y Rico, F., letter (Dec. 1956).
24. Catalán, M. A. y Rico, F., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **53**, Nos. 3, 4, 85-94 (1957).
25. Catalán, M. A. y Rico, F. R., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **54**, Nos. 9, 10, 289-300 (1958).
26. Catalán, M. A. y Rico, F., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **54**, Nos. 9, 10, 301-308 (1958).
27. Catalán, M. A. y Velasco, R., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **48**, Nos. 9, 10, 247-266 (1952).
28. Chaghtai, M. S.-U.-Z., *J. Opt. Soc. Am.* **59**, No. 8, 969-970 (1969).
29. Charles, G. W., *J. Opt. Soc. Am.* **56**, No. 10, 1292-1297 (1966).
30. Cohen, L., Feldman, U., and Kastner, S. O., *J. Opt. Soc. Am.* **58**, No. 3, 331-334 (1968).
31. Conway, J. G., and Worden, E. F., *J. Opt. Soc. Am.* **58**, No. 11, 1564 (A) (1968).
32. Cowan, R. D., *Astroph. J.* **147**, No. 1, 377-379 (L) (1967).
33. Cowan, R. D., *J. Opt. Soc. Am.* **58**, No. 7, 924-933 (1968), and Edlén, B., unpublished material (1970).
34. Cowan, R. D., and Peacock, N. J., *Astroph. J.* **142**, No. 1, 390-396 (L) (1965).
35. Crooker, A. M., private communication (December 1955).
36. Crooker, A. M., unpublished analysis (1962).
37. Crooker, A. M., and Dick, K. A., *Canadian J. Phys.* **46**, No. 10, 1241-1251 (1968).
38. Crooker, A. M., and Joshi, Y. N., *J. Opt. Soc. Am.* **54**, No. 4, 553-554 (L) (1964).
39. Deutschman, W. A., and House, L. L., *Astroph. J.* **144**, No. 1, 435-437 (L) (1966).
40. Deutschman, W. A., and House, L. L., *Astroph. J.* **149**, No. 2, Part 1, 451-452 (1967).
41. Dick, K. A., *Canadian J. Phys.* **46**, No. 11, 1291-1302 (1968).
42. Edlén, B., *Mon. Not. Roy. Astron. Soc.* **114**, No. 6, 700-703 (1954).
43. Edlén, B., *Trans. Intern. Astron. Union* **9**, 218-226 (1955).
44. Edlén, B., "Atomic Spectra," *Hand. der Phys., Encycl. of Phys.* **27**, 172 (1964).
45. Edlén, B., "Atomic Spectra," *Hand. der Phys., Encycl. of Phys.* **27**, 196 (1964).
46. Edlén, B., "Atomic Spectra," *Hand. der Phys., Encycl. of Phys.* **27**, 198 (1964).
47. Edlén, B., "Atomic Spectra," *Hand. der Phys., Encycl. of Phys.* **27**, 200 (1964).

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TABLE 2. Ionization limits and lowest terms—continued

[illegible]

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum																			
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	XX
21	Sc	3d 4s ² ² D _{3/2} 52750 ² D _{5/2} 168.34 27, 135	3d 4s ² D _{3/2} 103240 ² D _{5/2} 67.68 ² D _{3/2} 177.63 ² D _{5/2} 2540.97 135, 166	3d ² D _{3/2} 199700 ² D _{5/2} 197.5 25, 135	3p ² ¹ S ₀ 592600 52	3p ² ³ P _{1/2} 739300 ³ P _{3/2} 33.48 ³ P _{5/2} 4318 52, 182	3p ² ¹ P _{1/2} 896000 ¹ P _{3/2} 33.48 ¹ P _{5/2} 4458 ¹ D ₂ 21398 135, 182	3p ² ³ S ₁ 1113100 ³ S ₂ 29565 ³ S ₃ 30247 ³ D ₂ 25035 52, 116	3p ² ³ P ₀ 1280000 ³ P ₁ 2272 ³ P ₂ 5505 ³ D ₂ 25035 52, 60, 135	3p ² P _{3/2} 1452000 ² P _{1/2} 5761 52	3s ² ¹ S ₀ 1817400 52	3s ² S _{1/2} 2015080 52	2p ² ¹ S ₀ 5532200 46	2p ² ³ P ₀ 6093400 ³ P ₁ 37900 46, 59	2p ² ³ P ₂ 6692900 ³ P ₁ ³ P ₀ 46	2p ² ¹ S ₀ 7468900 46					
22	Ti	3d ² 4s ² ³ F ₂ 55010 ³ F ₃ 170.132 ³ F ₄ 386.874 27, 73	3d ² 4s ⁴ F _{7/2} 109506 ⁴ F _{9/2} 93.94 ⁴ F _{5/2} 225.47 ⁴ F _{3/2} 393.22 135, 166	3d ² ³ F ₂ 221735 ³ F ₃ 184.9 ³ F ₄ 420.4 ³ D ₂ 8473.5 ³ D ₃ 10538.4 ³ P ₁ 10603.6 ³ P ₂ 10721.2 51	3d ³ D _{3/2} 348973 ³ D _{5/2} 382.1 51	3p ² ¹ S ₀ 800300 52	3p ² ³ P _{1/2} 962700 ³ P _{3/2} 5829 52, 182	3p ² ³ P ₀ 1136000 ³ P ₁ 4530 ³ P ₂ 5884 ³ D ₂ 24130 135, 182	3p ² ³ S ₁ 1359000 ³ D _{1/2} 32168 ³ D _{3/2} 33239 33, 52	3p ² ³ P ₀ [1558000] ³ P ₁ 3125 ³ P ₂ 7291 ³ D ₂ 28557 52, 116, 183	3p ² P _{3/2} 1741500 ² P _{1/2} 7542 52	3s ² ¹ S ₀ 2139300 52	3s ² S _{1/2} 2351140 52	2p ² ¹ S ₀ 6350400 46	2p ² ³ P ₀ 6947300 ³ P ₁ 47190 46, 59	2p ² ³ P ₂ 7584700 ³ P ₁ ³ P ₀ 46					
23	V	3d ³ 4s ² ⁴ F _{7/2} 54400 ⁴ F _{9/2} 137.38 ⁴ F _{5/2} 323.42 ⁴ F _{3/2} 553.02 27, 135	3d ³ ⁴ D ₀ 118200 ⁴ D ₂ 36.05 ⁴ D ₄ 106.63 ⁴ D ₃ 208.69 ⁴ D ₄ 339.21 135, 188	3d ³ ⁴ F _{7/2} 236 ⁴ F _{9/2} 145.5 ⁴ F _{5/2} 341.5 ⁴ F _{3/2} 583.8 83	3d ² ³ F ₂ 376730 ³ F ₃ ± 40 ³ F ₄ 325.4 ³ F ₁ 734.7 ³ D ₂ 10959.3 85	3d ³ D _{3/2} [526100] ³ D _{5/2} 620 85, 116, 135	3p ² ¹ S ₀ 1033400 52	3p ² ³ P _{1/2} 1211200 ³ P _{3/2} 7660 52, 135	3p ² ³ P ₀ 1401000 ³ P ₁ 6000 ³ P ₂ 7580 ³ D ₂ 27120 135	3p ² ³ S ₁ [1659900] 116	3p ² ³ P ₀ [1859200] ³ P ₁ 4310 ³ P ₂ 9540 60, 116	3p ² P _{3/2} 2057100 ² P _{1/2} 9692 52	3s ² ¹ S ₀ 2486300 52	3s ² S _{1/2} 2712250 52	2p ² ¹ S ₀ 7223500 46	2p ² ³ P ₀ 7856200 ³ P ₁ ³ P ₂ 30, 46					
24	Cr	3d ² 4s ¹ S ₂ 54570 103	3d ² ³ S _{3/2} 133060 102	3d ² ³ D ₀ 249700 ³ D ₁ 59.9 ³ D ₂ 181.9 ³ D ₃ 355.8 ³ D ₄ 575.0 136	3d ² ³ F _{1/2} [395000] ³ F _{3/2} 244 ³ F _{5/2} 561 ³ F ₄ 956 116, 136	3d ² ³ F ₂ [559000] ³ F ₃ 513 ³ F ₄ 1146 ³ D ₂ 13200 116, 136	3d ³ D _{3/2} 730400 ³ D _{5/2} 920 3	3p ² ¹ S ₀ 1299700 136	3p ² ³ P _{1/2} [1490000] ³ P _{3/2} 9900 116, 136	3p ² ³ P ₀ [1688000] ³ P ₁ 7860 ³ P ₂ 9600 116, 136	3p ² ³ S ₁ [1971000] 116	3p ² ³ P ₀ [2184000] ³ P ₁ ³ P ₂ 116	3p ² P _{3/2} [2403600] ² P _{1/2} 12200 60, 116	3s ² ¹ S ₀ 2862000 136	3s ² S _{3/2} 3099630 136	2p ² ¹ S ₀ 8151600 46					
25	Mn	3d ² 4s ² ⁶ S _{5/2} 59970 21	3d ² 4s ¹ S ₂ 126145.0 ± 0.6 86	3d ² ⁶ S _{5/2} 271550 69	3d ² ⁵ D ₀ [413000] ⁵ D ₁ 98.4 ⁵ D ₂ 286.8 ⁵ D ₃ 552.7 ⁵ D ₄ 885.4 116, 138	3d ² ⁴ F _{7/2} [584000] ⁴ F _{9/2} 349 ⁴ F _{5/2} 827 ⁴ F _{3/2} 1406 116, 136	3d ² ³ F ₂ [766000] ³ F ₃ 746 ³ F ₄ 1669 ³ D ₂ 15337 17, 116, 136	3d ³ D _{3/2} 962001 ³ D _{5/2} 1355 136	3p ² ¹ S ₀ [1584600] 116	3p ² ³ P _{1/2} [1789000] ³ P _{3/2} 12530 116, 136	3p ² ³ P ₀ [2003000] ³ P ₁ 10000 ³ P ₂ 11700 116, 136	3p ² ³ S ₁ [2307000] 116	3p ² ³ P ₀ [2535900] ³ P ₁ ³ P ₂ 116	3p ² P _{3/2} [2771400] ² P _{1/2} 16400 62, 116	3s ² ¹ S ₀ 3260000 136	3s ² S _{3/2} 3511210 136	2p ² ¹ S ₀ 9164300 136				
26	Fe	3d ⁶ 4s ² ⁵ D ₄ 63480 ⁵ D ₂ 415.932 ⁵ D ₂ 704.003 ⁵ D ₁ 888.132 ⁵ D ₀ 978.076 27, 43	3d ⁶ 4s ⁴ D ₀ 130524 ⁴ D ₂ 384.77 ⁴ D ₄ 667.64 ⁴ D ₃ 862.63 ⁴ D ₀ 977.03 136, 166	3d ⁶ ³ D ₁ 247221 ³ D ₂ 436.2 ³ D ₃ 738.9 ³ D ₄ 932.4 ³ D ₅ 1027.3 74, 136	3d ² ⁶ S _{5/2} [442000] 116	3d ² ⁵ D ₀ [604900] ⁵ D ₁ 145 ⁵ D ₂ 419 ⁵ D ₃ 804 ⁵ D ₄ 1285 116, 136	3d ² ⁴ F _{7/2} [798500] ⁴ F _{9/2} 510.9 ⁴ F _{5/2} 1189 ⁴ F _{3/2} 2002 17, 116	3d ² ³ F ₂ [1008000] ³ F ₃ 1051 ³ F ₄ 2327 17, 116	3d ³ D _{3/2} 1218400 ³ D _{5/2} 1850 3, 32, 34	3p ² ¹ S ₀ 1895800 3	3p ² ³ P _{1/2} 2114000 ³ P _{3/2} 15682.9 89, 136	3p ² ³ P ₀ 2342000 ³ P ₁ 12667.9 ³ P ₂ 14440 89, 136	3p ² ³ S ₁ [2668000] 116	3p ² ³ P ₀ [2912000] ³ P ₁ 9802.6 ³ P ₂ 18561.1 89, 116	3p ² P _{3/2} [3163000] ² P _{1/2} 18850.6 89, 116	3s ² ¹ S ₀ 3684000 136	3s ² S _{3/2} 3947840 136	2p ² ¹ S ₀ 10211700 136			
27	Co	3d ⁷ 4s ² ⁴ F _{9/2} 63430 ⁴ F _{7/2} 816.00 ⁴ F _{5/2} 1406.84 ⁴ F _{3/2} 1809.33 27, 136	3d ⁷ ³ F ₄ 137572 ³ F ₂ 950.45 ³ F ₂ 1597.20 166, 187	3d ⁷ ⁴ F _{9/2} 270200 ⁴ F _{7/2} 841.2 ⁴ F _{5/2} 1451.3 ⁴ F _{3/2} 1866.8 171	3d ² ⁵ D ₀ [413800] 116	3d ² ⁵ D ₁ [609000] ⁵ D ₂ 208 ⁵ D ₃ 586 ⁵ D ₄ 1129 ⁵ D ₅ 1789 116, 136	3d ² ⁴ F _{7/2} [1040000] ⁴ F _{9/2} 698 ⁴ F _{5/2} 1610 ⁴ F _{3/2} 2723 116, 136	3d ² ³ F ₂ [1266000] ³ F ₃ 1430 ³ F ₄ 3140 4, 116	3d ³ D _{3/2} 1501300 ³ D _{5/2} 2440 3, 75	3p ² ¹ S ₀ 2230000 3	3p ² ³ P _{1/2} 2462000 ³ P _{3/2} 19420 75, 136	3p ² ³ P ₀ [2710000] ³ P ₁ ³ P ₂ 116	3p ² ³ S ₁ [3057000] 116	3p ² ³ P ₀ [3315000] ³ P ₁ [11834] ³ P ₂ 116, 136	3p ² P _{3/2} [3581000] ² P _{1/2} ² P ₂ 116	3s ² ¹ S ₀ 4133000 136	3s ² S _{3/2} 4410480 136	2p ² ¹ S ₀ 11316400 136			
28	Ni	3d ⁸ 4s ² ³ F ₄ 61579 ³ F ₂ 1332.153 ³ F ₂ 2216.519 136	3d ⁸ ¹ D _{3/2} 146541.56 ¹ D _{5/2} 1506.94 174	3d ⁸ ³ F ₄ 283700 ³ F ₂ 1188.4 ³ F ₂ 2040.9 ³ F ₂ 2618.8 27, 170	3d ² ⁶ S _{5/2} [442800] ⁶ S _{3/2} 1188.4 ⁶ S _{3/2} 2040.9 ⁶ S _{3/2} 2618.8 116, 148	3d ² ⁵ D ₀ [609000] ⁵ D ₁ 275 ⁵ D ₂ 802 ⁵ D ₃ 1520 ⁵ D ₄ 2379 116, 136	3d ² ⁴ F _{7/2} [1307000] ⁴ F _{9/2} 1890 ⁴ F _{5/2} 4110 ⁴ F _{3/2} 116	3d ² ³ F ₂ [1557000] ³ F ₃ 1890 ³ F ₄ 4110 58, 116	3d ³ D _{3/2} 1811000 ³ D _{5/2} 3150 3, 67	3p ² ¹ S ₀ 2591000 3	3p ² ³ P _{1/2} [2839000] ³ P _{3/2} 23627.3 89, 116	3p ² ³ P ₀ [3097000] ³ P ₁ 19541.8 ³ P ₂ [20200] 89, 116, 136	3p ² ³ S ₁ [3468000] 116	3p ² ³ P ₀ [3742500] ³ P ₁ 14917.5 ³ P ₂ 27376.5 89, 116	3p ² P _{3/2} [4025000] ² P _{1/2} 27761.4 ² P ₂ 89, 116	3s ² ¹ S ₀ [4606000] 116	3s ² S _{3/2} 4897400 136	2p ² ¹ S ₀ 12477000 61			
29	Cu	4s ² S _{1/2} 62317.2 136	3d ¹⁰ ¹ S ₀ 163669.2 ± 0.5 163	3d ⁹ ² D _{3/2} 297100 ² D _{5/2} 2071.8 136	3d ⁸ ³ F ₄ 445124 ± 1600 ³ F ₂ 1862.5 ³ F ₂ 3078.5 169	3d ⁷ ⁴ F _{9/2} [644500] ⁴ F _{7/2} ⁴ F _{5/2} ⁴ F _{3/2} 116	3d ⁶ ³ D ₁ [831000] ³ D ₂ ³ D ₃ ³ D ₄ 116	3d ⁵ ⁴ S _{3/2} [1121000] 116	3d ⁴ ³ D ₀ [1339000] ³ D ₁ ³ D ₂ ³ D ₃ ³ D ₄ 116	3d ³ ⁴ F _{7/2} [1605000] ⁴ F _{9/2} ⁴ F _{5/2} ⁴ F											

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum																				
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	XX	XXI
30	Zn	4s ² 95 ¹ S ₀ 75768.10	4s 122 ² S _{1/2} 144892.6 ± 2	3d ¹⁰ 41 ¹ S ₀ 320390 ± 1	3d ⁸ 37, 116 ³ D _{3/2} [479100] ¹ D ₂ 2758.8	3d ⁸ 116 ³ F ₂ [666000] ³ F ₃ ³ F ₄	3d ⁷ 116 ⁴ F _{3/2} [871000] ⁴ F _{5/2} ⁴ F _{7/2} ⁴ F _{9/2}	3d ⁸ 116 ³ D ₃ [1081000] ³ D ₅ ³ D ₁ ³ D ₀	3d ⁷ 116 ³ S _{1/2} [1403000]	3d ⁸ 116 ³ D ₃ [1637000] ³ D ₅ ³ D ₁ ³ D ₀	3d ⁸ 116 ⁴ F _{3/2} [1920000] ⁴ F _{5/2} ⁴ F _{7/2} ⁴ F _{9/2}	3d ⁷ 58, 116 ³ F ₂ [2210000] ³ F ₃ 3030 ³ F ₄ 6735	3d 116 ³ D _{3/2} [2507000]	3p ² 116 ³ S ₀ [3385000]	3p ² 116 ³ P _{1/2} [3662000] ³ P _{3/2}	3p ⁴ 116 ³ P ₂ [3952000] ³ P ₁ ³ P ₀	3p ³ 116 ⁴ S _{3/2} [4372000]	3p ² 116 ³ P ₂ [4670000] ³ P ₁ ³ P ₀	3p 116 ³ P _{1/2} [4993000] ³ P _{3/2}	3s ² 116 ³ S ₀ [5630000]	3s 64 ³ S _{1/2} 5952000	2p ⁴ 63 ³ S ₀ 14960000
31	Ga	4p 96 ² P _{1/2} 48387.63 ² P _{3/2} 826.19	4s ² 130 ¹ S ₀ 165458	4s 136 ² S _{1/2} 247700	3d ¹⁰ 136 ¹ S ₀ 517600																	
32	Ge	4p ² 6, 97 ³ P ₀ 63715 ± 10 ³ P ₁ 557.1341 ³ P ₂ 1409.9609 ¹ D ₂ 7125.2989 ¹ S ₀ 16367.3332	4p 97, 100 ³ P ₀ 128521.3 ± 0.2 ³ P ₁ 1767.356	4s ² 136 ¹ S ₀ 276036	4s 136 ² S _{1/2} 368701	3d ¹⁰ 136 ¹ S ₀ 753800																
33	As	4p ³ 136 ² S _{1/2} 79165	4p ² 35 ³ P ₀ 150290 ³ P ₁ 1061 ³ P ₂ 2538 ¹ D ₂ 10097 ¹ S ₀ 22602	4p 35, 136 ³ P ₀ 228670 ³ P ₁ 2940	4s ² 136 ¹ S ₀ 404369	4s 136 ² S _{1/2} 505136	3d ¹⁰ 136 ¹ S ₀ 1028800															
34	Se	4p ⁴ 136 ³ P ₂ 78658.22 ³ P ₁ 1989.49 ³ P ₀ 2534.35 ¹ D ₂ 9576.08 ¹ S ₀ 22446.03	4p ³ 26 ² S _{1/2} 170900	4p ² 36, 136 ³ P ₀ 248583 ³ P ₁ 1741 ³ P ₂ 3937 ¹ D ₂ 13032	4s ² 65 ¹ S ₀ [551000]	4s 136 ² S _{1/2} 658994	3d ¹⁰ 136 ¹ S ₀ 1253300															
35	Br	4p ⁵ 184 ² P _{1/2} 95284.8 ± 0.5 ² P _{3/2} 3685.24	4p ⁴ 152, 184 ³ P ₂ 175870 ³ P ₁ 3136.4 ³ P ₀ 3837.5 ¹ D ₂ 12089.1 ¹ S ₀ 27867.1	4p ³ 136 ² S _{1/2} 289529	4p ² 65, 136 ³ P ₀ [381600] ³ P ₁ 3247 ³ P ₂ 6237 ¹ D ₂ 18115	4p 65, 136 ² P _{1/2} [481600] ² P _{3/2} 6090	4s ² 65 ¹ S ₀ [714800]	4s 65 ² S _{1/2} [831000]	3d ¹⁰ 136 ¹ S ₀ 1554700													
36	Kr	4p ⁶ 136, 147 ¹ S ₀ 112914.5	4p ⁵ 132, 136 ² P _{1/2} 196474.8 ² P _{3/2} 5371.00	4p ⁴ 136 ³ P ₂ 298020 ³ P ₁ 4548 ³ P ₀ 5313 ¹ D ₂ 14644 ¹ S ₀ 33079	4p ³ 65 ² S _{1/2} [423600]	4p ² 65 ³ P ₀ [522000] ³ P ₁ ³ P ₂	4p 61, 65 ² P _{1/2} [633300] ² P _{3/2} 8108	4s ² 65 ¹ S ₀ [895500]	1s 65 ² S _{1/2} [1016500]	3d ¹⁰ 106 ¹ S ₀ 1862400												
37	Rb	5s 91 ² S _{1/2} 33690.81 ± 0.01	4p ⁶ 155a ¹ S ₀ 220048 ± 30	4p ⁵ 136 ² P _{1/2} 320000 ² P _{3/2} 7380	4p ⁴ 65 ³ P ₂ [424400] ³ P ₁ ³ P ₀	4p ³ 65 ² S _{1/2} [572800]	4p ² 65 ³ P ₂ [680900] ³ P ₁ ³ P ₀	4p 65 ² P _{1/2} [800300] ² P _{3/2}	4s ² 65 ¹ S ₀ [1098000]	4s 65 ² S _{1/2} [1210000]	3d ¹⁰ 136 ¹ S ₀ 2235100											
38	Sr	5s ² 70 ¹ S ₀ 45932.0 ± 0.2	5s 136 ² S _{1/2} 88964.0	4p ⁶ 65 ¹ S ₀ [351800]	4p ⁵ 136 ² P _{1/2} 460000 ² P _{3/2} 9731	4p ⁴ 65 ³ P ₂ [577700]	4p ³ 65 ² S _{1/2} [732600]	4p ² 65 ³ P ₂ [855200] ³ P ₁ ³ P ₀	4p 65 ² P _{1/2} [986700] ² P _{3/2}	4s ² 65 ¹ S ₀ [1307000]	1s 65 ² S _{1/2} [1428000]	3d ¹⁰ 136 ¹ S ₀ 2613800										
39	Y	4d 5s ² 22, 136 ¹ D _{3/2} 51447 ¹ D _{5/2} 530.36	5s ² 22 ¹ S ₀ 98690	5s 24, 104 ² S _{1/2} 165500	4p ⁶ 65 ¹ S ₀ [498600]	4p ⁵ 65, 155 ² P _{1/2} [621200] ² P _{3/2} 12459.9	4p ⁴ 65 ³ P ₂ [750300] ³ P ₁ ³ P ₀	4p ³ 65 ² S _{1/2} [935900]	4p ² 65 ³ P ₀ [1011000]	4p 65 ² P _{1/2} [1179500] ² P _{3/2}	4s ² 65 ¹ S ₀ [1541000]	4s 65 ² S _{1/2} [1662000]	3d ¹⁰ 136 ¹ S ₀ 3016800									
40	Zr	4d ² 5s ² 22, 136 ³ F ₂ 55145 ³ F ₃ 570.41 ³ F ₄ 1240.84	4d ² 5s 23, 136 ³ F ₂ 105900 ³ F ₃ 314.67 ³ F ₄ 763.44 ¹ F _{3/2} 1322.91	4d ² 24, 104 ³ F ₂ 185400 ³ F ₃ 680.5 ³ F ₄ 1485.7	4d 104 ¹ D _{3/2} 276970 ¹ D _{5/2} 1250	4p ⁶ 28 ¹ S ₀ 657600																

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum								
		I	II	III	IV	V	VI	VII	VIII	IX
41	Nb	4d ⁴ 5s ⁴ D _{3/2} 5551.1 ⁴ D _{1/2} 154.19 ⁴ D _{5/2} 391.99 ⁴ D _{3/2} 695.25 ⁴ D _{1/2} 1050.26 22	4d ⁴ ¹ D ₀ 115500 ³ D ₁ 158.99 ³ D ₂ 438.38 ³ D ₃ 801.38 ³ D ₄ 1224.87 23, 136	4d ³ ⁴ F _{4/2} 202000 ⁴ F _{2/2} 515.8 ⁴ F _{3/2} 1176.6 ⁴ F _{4/2} 1939.0 24, 82	4d ² ³ F ₂ 308600 ³ F ₃ 1086.4 ³ F ₄ 2344.6 136	4d ¹ D _{3/2} 407700 ³ D _{3/2} 1870 104, 136	4p ⁴ ¹ S ₀ 827300 28	4p ² ¹ P _{1/2} 1005000 ³ P _{0/2} 19199 28, 136		
42	Mo	4d ⁵ 5s ¹ S ₀ 57260 137	4d ⁵ ⁴ S _{3/2} 130300 137	4d ⁴ ¹ D ₀ 219100 ³ D ₁ 243.10 ³ D ₂ 669.60 ³ D ₃ 1225.20 ³ D ₄ 1873.80 158	4d ³ ⁴ F _{4/2} 374180 ⁴ F _{2/2} 780.0 ⁴ F _{3/2} 1759.0 ⁴ F _{4/2} 2858.6 137	4d ² ³ F ₂ 493360 ³ F ₃ 1585 ³ F ₄ 3359 137	4d ¹ D _{3/2} 549000 ³ D _{3/2} 2578 104, 137	4p ⁴ ¹ S ₀ 1022800 28	4p ² ¹ P _{1/2} 1235000 ³ P _{0/2} 23273 28, 137	
43	Tc	4d ⁵ 5s ² ⁴ S _{3/2} 58700 137	4d ⁵ 5s ¹ S ₂ 123100 137	4d ⁴ ⁴ S _{3/2} 238300 24						
44	Ru	4d ⁶ 5s ³ F ₄ 59410 ³ F ₄ 1190.64 ³ F ₃ 2091.54 ³ F ₂ 2713.24 ³ F ₁ 3105.49 101	4d ⁷ ⁴ F _{4/2} 135200 ⁴ F _{3/2} 1523.1 ⁴ F _{3/2} 2493.9 ⁴ F _{4/2} 3104.2 175	4d ⁶ ¹ D ₄ 229600 ³ D ₃ 1158.8 ³ D ₂ 1826.3 ³ D ₁ 2266.3 ³ D ₀ 2476.0 137						
45	Rh	4d ⁸ 5s ⁴ F _{4/2} 60197 ⁴ F _{3/2} 1529.97 ⁴ F _{3/2} 2598.03 ⁴ F _{4/2} 3472.68 137	4d ⁸ ³ F ₄ 145800 ³ F ₃ 2401.3 ³ F ₂ 3580.7 137	4d ⁷ ⁴ F _{4/2} 250500 ⁴ F _{3/2} 2147.8 ⁴ F _{3/2} 3485.7 ⁴ F _{4/2} 4322.0 84, 137						
46	Pd	4d ¹⁰ 137	¹ S ₀ 67236.0 137	¹ D _{3/2} 156700 ³ D _{3/2} 3539 137	4d ⁸ ³ F ₄ 265600 ³ F ₃ 3229.3 ³ F ₂ 4687.5 173					
47	Ag	5s 137	¹ S ₀ 61106.50 137	4d ¹⁰ ¹ S ₀ 173300 137	4d ⁹ ³ D _{3/2} 280900 ³ D _{5/2} 4607 137					
48	Cd	5s ² 137	¹ S ₀ 72538.8 137	5s ¹ S ₀ 136374.74 137	4d ¹⁰ ¹ S ₀ 302300 137					
49	In	5s ² 5p ² P _{1/2} 46670.11 ± 0.05 ² P _{3/2} 2212.598 96	5s ² ¹ S ₀ 152195 137	5s ¹ S ₀ 226100 137	4d ¹⁰ ¹ S ₀ 439000 137					
50	Sn	5p ² ¹ P _{1/2} 59231.8 ³ P ₁ 1691.8 ³ P ₂ 3427.7 137	5s ² 5p ² P _{1/2} 118017.0 ² P _{3/2} 4251.4 137	5s ² ¹ S ₀ 246020.0 137	5s ¹ S ₀ 328550.0 137	4d ¹⁰ ¹ S ₀ 583000 137				
51	Sb	5p ³ ⁴ S _{3/2} 69700 137	5p ³ ³ P ₀ 133327.5 ³ P ₁ 3055.0 ³ P ₂ 5659.0 137	5s ² 5p ² P _{1/2} 204248 ² P _{3/2} 6576 137	5s ² ¹ S ₀ 356156 137	5s ¹ S ₀ 449300 137	4d ¹⁰ ¹ S ₀ 868000 137			
52	Te	5p ⁴ ¹ P _{1/2} 72667 ³ P ₁ 4751 ³ P ₀ 4707 137	5p ⁴ ⁴ S _{3/2} 150000 ± 3000 79	5p ³ ¹ P ₀ 225500 ³ P ₁ 4756.5 ³ P ₂ 8166.9 38	5s ² 5p ² P _{1/2} 301776 ² P _{3/2} 9222.6 38	5s ² ¹ S ₀ 473900 38	5s ¹ S ₀ 570000 38	4d ¹⁰ ¹ S ₀ 1107000 137		
53	I	5p ⁵ ² P _{1/2} 84295.1 ± 0.2 ² P _{3/2} 7603.15 131	5p ⁴ ¹ P ₂ 154304 ± 1 ³ P ₁ 7087.0 ³ P ₀ 6447.9 121	5p ³ ⁴ S _{3/2} [266000] 65						

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
54	Xe	5p ⁴ 137, 147 ¹ S ₀ 97834.0	5p ³ 137 ² P ^o _{1/2} 171068.4 ² P ^o _{3/2} 10537.01	5p ⁴ 137 ³ P ₂ 259089 ³ P ₁ 9794.6 ³ P ₀ 8131			
55	Cs	6s 11 ² S _{0+1/2} 31406.432 ± 0.010	5p ³ 137 ¹ S ₀ 202263				
56	Ba	6s ² 71 ¹ S ₀ 42035.14 ± 0.05	6s 137 ² S _{0+1/2} 80686.87				
57	La	5d 6s ² 72, 137 ² D _{3/2} 44981 ± 5 ² D _{5/2} 1053.20	5d ² 6s ² 181 ³ F ₂ 89200 ± 650 ³ F ₃ 1016.10 ³ F ₄ 1970.70	5d 6s ² 139 ³ D _{3/2} 154664 ± 15 ³ D _{5/2} 1603.26			
58	Ce	4f 5d 6s ² 120, 156a ¹ G ^o _{7/2} 44090 ± 110	4f 5d ² 181 ⁴ H ^o _{11/2} 87500 ± 650 ⁴ H ^o _{9/2} 987.62 ⁴ H ^o _{7/2} 1873.95 ⁴ H ^o _{5/2} 2382.26	4f ² 178 ³ H ₄ 162900 ± 120 ³ H ₃ 1526.36 ³ H ₂ 3127.05	4f 113 ³ F ^o _{3/2} 296200 ³ F ^o _{5/2} 2253		
59	Pr	4f ³ 156, 193 6s ² 41a _{1/2} 43730 ± 150 41 _{3/2} 1376.54 41 _{5/2} 2846.61 41 _{7/2}	4f ³ 6s 181 ⁴ F ^o _{7/2} 85100 ± 650 ⁴ F ^o _{5/2} 441.94 ⁴ F ^o _{3/2} 1649.01 ⁴ F ^o _{1/2} 2998.31 ⁴ F ^o _{9/2} 4437.09	4f ³ 177, 180 41 _{1/2} 174420 ± 130 41 _{3/2} 1398.34 41 _{5/2} 2893.14 41 _{7/2} 4453.76	4f ² 179 ³ H ₄ 314200 ± 100 ³ H ₃ 2152.2 ³ H ₂ 4389.1	4f 99 ² F ^o _{3/2} 463400 ± 400 ² F ^o _{5/2} 3027.4	
60	Nd	4f ⁴ 156, 190 6s ² 1a 44270 ± 150 1b 1128.055 1c 2366.595 1d 3681.690 1e 5048.665	4f ⁴ 6s 181, 190 ⁴ I ^o _{13/2} 86500 ± 650 ⁴ I ^o _{11/2} 513.330 ⁴ I ^o _{9/2} 1470.100 ⁴ I ^o _{7/2} 2585.460 ⁴ I ^o _{5/2} 3801.935 ⁴ I ^o _{3/2} 5085.650				
61	Pm	4f ⁵ 154, 156 6s ² 4H ^o _{11/2} 44800 ± 150 4H ^o _{9/2} 803.82 4H ^o _{7/2} 1748.78 4H ^o _{5/2} 2797.10 4H ^o _{3/2} 3919.03 4H ^o _{1/2} 5089.79	4f ⁵ 6s 153, 181 ⁴ I ^o _{13/2} 87900 ± 650 ⁴ I ^o _{11/2} 446.45 ⁴ I ^o _{9/2} 1133.45 ⁴ I ^o _{7/2} 1983.52 ⁴ I ^o _{5/2} 2950.31 ⁴ I ^o _{3/2} ⁴ I ^o _{1/2}				
62	Sm	4f ⁶ 2, 156 6s ² 1F ^o ₀ 45420 ± 150 1F ^o ₁ 292.58 1F ^o ₂ 811.92 1F ^o ₃ 1489.55 1F ^o ₄ 2273.09 1F ^o ₅ 3125.46 1F ^o ₆ 4020.66	4f ⁶ 6s 1, 181 ⁶ F ^o _{0/4} 89300 ± 650 ⁶ F ^o _{1/2} 326.64 ⁶ F ^o _{3/2} 838.22 ⁶ F ^o _{5/2} 1489.16 ⁶ F ^o _{7/2} 2237.97 ⁶ F ^o _{9/2} 3052.65 ⁶ F ^o _{11/2} 3909.62				
63	Eu	4f ⁷ 168 6s ² 8S ^o _{7/2} 45740 ± 80	4f ⁷ 6s 167, 181 ⁸ S ^o _{7/2} 90700				

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
64	Gd	$4f^7 5d 6s^2 {}^6D_2^o$ 49530 ± 110 ${}^6D_3^o$ 215.13 ${}^6D_1^o$ 532.98 ${}^6D_5^o$ 999.11 ${}^6D_4^o$ 1719.06 156a, 165	$4f^7 5d 6s {}^{10}D_{5/2}^o$ 97900 ± 3000 ${}^{10}D_{5/2}^o$ 261.81 ${}^{10}D_{7/2}^o$ 633.27 ${}^{10}D_{3/2}^o$ 1158.94 ${}^{10}D_{1/2}^o$ 1935.30 165, 181				
65	Tb	$4f^9 6s^2 {}^4H_{7/2}^o$ [47200] ± 150 ${}^4H_{5/2}^o$ ${}^4H_{3/2}^o$ ${}^4H_{1/2}^o$ ${}^4H_{7/2}^o$ ${}^4H_{5/2}^o$ 156, 176	$4f^9 6s {}^7H_6$ 92900 ± 650 7H_7 7H_8 7H_9 7H_4 7H_5 7H_1 181				
66	Dy	$4f^{10} 6s^2 {}^1I_8$ 47820 ± 150 1I_7 4134.24 1I_6 7050.61 1I_5 1I_4 31, 156	$4f^{10} 6s {}^4I_{3/2}^o$ 94100 ± 650 ${}^4I_{7/2}^o$ 4341.10 ${}^4I_{5/2}^o$ 7485.09 ${}^4I_{3/2}^o$ 7463.88 ${}^4I_{1/2}^o$ 9432.07 ${}^4I_{5/2}^o$ 10953.94 31, 181				
67	Ho	$4f^{11} 6s^2 {}^4I_{7/2}^o$ 48540 ± 150 ${}^4I_{5/2}^o$ ${}^4I_{3/2}^o$ ${}^4I_{1/2}^o$ 156	$4f^{11} 6s {}^3I_5^o$ 95200 ± 650 ${}^3I_7^o$ ${}^3I_6^o$ ${}^3I_4^o$ ${}^3I_2^o$ 181				
68	Er	$4f^{12} 6s^2 {}^3H_8$ 49210 ± 150 3H_3 6958.34 3H_4 10750.99 117, 156	$4f^{12} 6s {}^4H_{5/2}^o$ 96200 ± 650 ${}^4H_{3/2}^o$ 7149.7 ${}^4H_{1/2}^o$ 11042.8 ${}^4H_{5/2}^o$ 10894.1 125, 181				
69	Tm	$4f^{13} 6s^2 {}^3F_{3/2}^o$ 49840 ± 150 ${}^3F_{5/2}^o$ 8771.25 126, 156	$4f^{13} 6s {}^3F_7$ 97200 ± 650 3F_3 236.94 3F_2 8769.69 126, 181	$4f^{13} {}^3F_{3/2}^o$ 191200 ± 500 ${}^3F_{5/2}^o$ 8774.02 180a			
70	Yb	$4f^{14} 6s^2 {}^1S_0$ 50441.0 ± 0.2 20a	$4f^{14} 6s {}^3S_{3/2}^o$ 98150 128	$4f^{14} {}^1S_0$ 203300 20			
71	Lu	$5d 6s^2 {}^1D_{3/2}^o$ 43762.39 ± 0.10 ${}^1D_{5/2}^o$ 1993.92 206, 110	$6s^2 {}^1S_0$ 112000 ± 3000 181				
72	Hf	$5d^2 6s^2 {}^3F_3$ 56600 3F_2 2356.68 3F_4 4567.64 127	$5d 6s^2 {}^3D_{3/2}^o$ 120000 ${}^3D_{5/2}^o$ 3050.83 137	$5d^2 {}^3F_2$ 187800 3F_2 3288.7 3F_4 6095.1 111	$5d {}^1D_{3/2}^o$ 268500 ± 800 ${}^1D_{5/2}^o$ 4692.0 111		
73	Ta	$5d^3 6s^2 {}^4F_{7/2}^o$ 63600 ${}^4F_{5/2}^o$ 2010.10 ${}^4F_{3/2}^o$ 3963.92 ${}^4F_{1/2}^o$ 5621.04 137					

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum											
		I		II		III		IV		V		VI	
74	W	5d ⁴ 6s ²	⁵ D ₀ 64400 ⁵ D ₁ 1670.29 ⁵ D ₂ 3325.53 ⁵ D ₃ 4830.00 ⁵ D ₄ 6219.33 114, 137										
75	Re	5d ⁵ 6s ²	⁶ S _{21/2} 63530 137										
76	Os	5d ⁶ 6s ²	⁵ D ₄ 70450 ⁵ D ₃ 4159.32 ⁵ D ₂ 2740.49 ⁵ D ₁ 5766.14 ⁵ D ₀ 6092.79 108, 137										
77	Ir	5d ⁷ 6s ²	⁴ F _{41/2} 73000 ± 800 ⁴ F _{31/2} 6323.96 ⁴ F _{21/2} 5784.63 ⁴ F _{11/2} 4078.95 107										
78	Pt	5d ⁹ 6s	³ D ₃ 72300 ³ D ₂ 775.9 ³ D ₁ 10132.0 137	5d ⁹ 137	² D _{21/2} 149723 ² D _{11/2} 8419.9								
79	Au	5d ¹⁰ 6s	² S _{01/2} 74410.0 137	5d ¹⁰ 137	¹ S ₀ 165000								
80	Hg	6s ²	¹ S ₀ 84184.1 137	5d ¹⁰ 6s 137	² S _{01/2} 151280	5d ¹⁰ 137	¹ S ₀ 276000						
81	Tl	6s ² 6p	² P _{01/2} ^o 49266.7 ± 0.1 157 ² P _{11/2} ^o 7792.7	6s ² 137	¹ S ₀ 164765 ± 5	6s 137	² S _{01/2} 240600						
82	Pb	6p ²	³ P ₀ 59819.4 ± 0.3 ³ P ₁ 7819.2626 ³ P ₂ 10650.3271 189	6s ² 6p 137, 189	² P _{01/2} ^o 121243 ± 3 ² P _{11/2} ^o 14081.074	6s ² 137	¹ S ₀ 257592 ± 5	6s 137	² S _{01/2} 341350	5d ¹⁰ ¹ S ₀ 137	555000		
83	Bi	6p ³	⁴ S _{11/2} ^o 58790 137	6p ² 137	³ P ₀ 134600 ³ P ₁ 13324 ³ P ₂ 17030	6s ² 6p 137	² P _{01/2} ^o 206180 ² P _{11/2} ^o 20788	6s ² 137	¹ S ₀ 365500	6s ² S _{01/2} 137	451700	5d ¹⁰ ¹ S ₀ 137	712000
84	Po	6p ⁴	³ P ₂ 67885.3 ³ P ₁ 16831.61 ³ P ₀ 7514.69 29										
85	At												
86	Rn	6p ⁶	¹ S ₀ 86692.5 137										
87	Fr												
88	Ra	7s ²	¹ S ₀ 42577.35 137	7s 137	² S _{01/2} 81842.31								
89	Ac	6d 7s ²	² D _{11/2} [55600] ² D _{21/2} 2231.43 65, 137	7s ² 137	¹ S ₀ 97300								

TABLE 2. Ionization limits and lowest terms—Continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
90	Th	$6d^2 7s^2$ 3F_2 3F_3 2869.260 3F_4 4961.661 192	$6d^2 7s$ $^4F_{11/2}$ [93000] $^4F_{21/2}$ 1521.91 $^4F_{31/2}$ 4146.57 $^4F_{41/2}$ 6213.55 65, 124	$6d^2$ 3F_2 161000 3F_3 3992.7 3F_4 6474.9 109	$5f$ $^2F_{21/2}^o$ 231900 $^2F_{31/2}^o$ 4325.38 112		
91	Pa						
92	U						
93	Np						
94	Pu	$5f^8 7s^2$ 7F_0 47000 7F_1 2203.55 7F_2 4299.55 7F_3 6144.34 7F_4 7774.45 7F_5 9179.05 7F_6 10238.24 7, 8					
95	Am	$5f^7 7s^2$ $^8S_{3/2}^o$ 48770 66					

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References—Continued

48. Edlén, B., *Solar Physics* **9**, No. 2, 439–445 (1969).
49. Edlén, B., unpublished material (1969), (1970).
50. Edlén, B., and Risberg, P., *Ark. Fys. (Stockholm)* **10**, No. 39, 553–566 (1956).
51. Edlén, B., and Swensson, J. W., unpublished material (Jan. 1970).
52. Ekberg, J. O., and Svensson, L. Å., unpublished material (Jan. 1970).
53. Eriksson, K. B. S., *Ark. Fys. (Stockholm)* **13**, No. 25, 303–329 (1958).
54. Eriksson, K. B. S., *Ark. Fys. (Stockholm)* **30**, No. 16, 199–202 (1965).
55. Eriksson, K. B. S., *Ark. Fys. (Stockholm)* **33**, No. 25, 357–360 (1966).
56. Eriksson, K. B. S., and Isberg, H. B. S., *Ark. Fys. (Stockholm)* **23**, No. 47, 527–542 (1963); **33**, No. 39, 593–595 (1966).
57. Eriksson, K. B. S., and Isberg, H. B. S., *Ark. Fys. (Stockholm)* **37**, No. 17, 221–230 (1968).
58. Even-Zohar, M., and Fraenkel, B. S., *J. Opt. Soc. Am.* **58**, No. 10, 1420–1421 (1968).
59. Fawcett, B. C., Burgess, D. D., and Peacock, N. J., *Proc. Phys. Soc. (London)* **91**, Part 4, No. 574, 970–972 (1967).
60. Fawcett, B. C., Gabriel, A. H., and Saunders, P. A. H., *Proc. Phys. Soc. (London)* **90**, Part 3, No. 569, 863–867 (1967).
61. Fawcett, B. C., Jones, B. B., and Wilson, R., *Proc. Phys. Soc. (London)* **78**, Part 6(i), No. 505, 1223–1226 (1961).
62. Fawcett, B. C., and Peacock, N. J., *Proc. Phys. Soc. (London)* **91**, Part 4, No. 574, 973–975 (1967).
63. Feldman, U., Cohen, L., and Swartz, M., *Astroph. J.* **148**, No. 2, 585–587 (1967).
64. Feldman, U., Cohen, L., and Swartz, M., *J. Opt. Soc. Am.* **57**, No. 4, 535–536 (1967).
65. Finkelburg, W., and Humbach, W., *Naturwiss.* **42**, 35–37 (1955).
66. Fred, M., and Tomkins, F. S., *J. Opt. Soc. Am.* **47**, No. 12, 1076–1087 (1957).
67. Gabriel, A. H., Fawcett, B. C., and Jordan, C., *Proc. Phys. Soc. (London)* **87**, Part 3, No. 557, 825–839 (1966).
68. Garcia, J. D., and Mack, J. E., *J. Opt. Soc. Am.* **55**, No. 6, 654–685 (1965).
69. Garcia-Riquelme, O., *Opt. Pura y Aplicada (Madrid)* **1**, No. 2, 53–72 (1968).
70. Garton, W. R. S., and Codling, K., *J. Phys. B (Proc. Phys. Soc. London)* [2] **1**, No. 1, 106–113 (1968).
71. Garton, W. R. S., and Tomkins, F. S., *Astroph. J.* **158**, No. 3, 1219–1230 (1969).
72. Garton, W. R. S., and Wilson, M., *Astroph. J.* **145**, No. 1, 333–336 (1966).
73. Giuliani, J. F., and Thekaekara, M. P., *J. Opt. Soc. Am.* **54**, No. 4, 460–463 (1964).
74. Glad, S., *Ark. Fys. (Stockholm)* **10**, No. 22, 291–340 (1956).
75. Goldsmith, S., *J. Opt. Soc. Am.* **59**, No. 12, 1678–1679 (L) (1969).
76. Hallin, R., *Ark. Fys. (Stockholm)* **31**, No. 36, 511–526 (1966).
77. Hallin, R., *Ark. Fys. (Stockholm)* **32**, No. 11, 201–210 (1966).
78. Hallin, R., unpublished material (1969).
79. Handrup, M. B., and Mack, J. E., *Physica* **30**, No. 7, 1245–1275 (1964).
80. Harrison, G. R., Albertson, W. E., and Hosford, N. F., *J. Opt. Soc. Am.* **31**, No. 6, 439–448 (1941).
81. Herzberg, G., and Moore, H. R., *Canadian J. Phys.* **37**, No. 11, 1293–1313 (1959).
82. Iglesias, L., *J. Opt. Soc. Am.* **45**, No. 10, 856–861 (1955).
83. Iglesias, L., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **58**, Nos. 7, 8, 191–222 (1962); *Opt. Pura y Aplicada (Madrid)* **2**, 132–145 (1969).
84. Iglesias, L., *Canadian J. Phys.* **44**, No. 4, 895–915 (1966).
85. Iglesias, L., *J. Res. Nat. Bur. Stand. (U.S.)*, **72A** (Phys. and Chem.) No. 4, 295–308 (July–Aug. 1968).
86. Iglesias, L., y Velasco, R., *Publ. Inst. Opt. (Madrid)* No. 23, 228 pp. (1964).
87. Isberg, B., *Ark. Fys. (Stockholm)* **35**, No. 45, 551–562 (1967).
88. Jakobsson, L. R., *Ark. Fys. (Stockholm)* **34**, No. 2, 19–31 (1967).
89. Jefferies, J. T., *Mem. Soc. Roy. des Sci. de Liège* **54**, Part 2, Section 3, 213–234 (1969).
90. Johansson, I., *Ark. Fys. (Stockholm)* **15**, No. 14, 169–179 (1959).
91. Johansson, I., *Ark. Fys. (Stockholm)* **20**, No. 7, 135–146 (1961).
92. Johansson, L., *Ark. Fys. (Stockholm)* **20**, No. 33, 489–498 (1961).
93. Johansson, L., *Ark. Fys. (Stockholm)* **23**, No. 12, 119–128 (1962).
94. Johansson, L., *Ark. Fys. (Stockholm)* **31**, No. 15, 201–235 (1966).
95. Johansson, I., and Contreras, R., *Ark. Fys. (Stockholm)* **37**, No. 31, 513–520 (1968).
96. Johansson, I., and Litzén, U., *Ark. Fys. (Stockholm)* **34**, No. 46, 573–587 (1967).
97. Kaufman, V., and Andrew, K. L., *J. Opt. Soc. Am.* **52**, No. 11, 1223–1237 (1962).
98. Kaufman, V., and Radziemski, L. J., Jr., *J. Opt. Soc. Am.* **59**, No. 2, 227–228 (L) (1969).
99. Kaufman, V., and Sugar, J., *J. Res. Nat. Bur. Stand. (U.S.)*, **71A** (Phys. and Chem.) No. 6, 583–585 (Nov.–Dec. 1967).
100. Kaufman, V., and Ward, J. F., *J. Opt. Soc. Am.* **56**, No. 11, 1591–1597 (1966).
101. Kessler, K. G., *J. Res. Nat. Bur. Stand. (U.S.)*, **63A** (Phys. and Chem.) No. 3, 213–251 (May–June 1959).
102. Kiess, C. C., *J. Res. Nat. Bur. Stand. (U.S.)* **47**, No. 5, 385–426 (1951), RP2266.
103. Kiess, C. C., *J. Res. Nat. Bur. Stand. (U.S.)* **51**, No. 5, 247–305 (1953), RP2457.
104. Kiess, C. C., *J. Res. Nat. Bur. Stand. (U.S.)* **56**, No. 4, 167–177 (1956), RP2663.
106. Kim, H. H., *J. Opt. Soc. Am.* **58**, No. 5, 739–740 (A) (1968).
107. van Kleef, Th. A. M., *Physica* **23**, No. 9, 843–897 (1957).
108. van Kleef, Th. A. M., and Klinkenberg, P. F. A., *Physica* **27**, No. 1, 83–94 (1961).
109. Klinkenberg, P. F. A., *Physica* **16**, Nos. 7, 8, 618–650 (1950).
110. Klinkenberg, P. F. A., *Physica* **21**, No. 1, 53–62 (1954).
111. Klinkenberg, P. F. A., van Kleef, Th. A. M., and Noorman, P. E., *Physica* **27**, No. 12, 1177–1188 (1961).
112. Klinkenberg, P. F. A., and Lang, R. J., *Physica* **15**, Nos. 8, 9, 774–788 (1949).
113. Lang, R. J., *Canadian J. Research* **A14**, 127–130 (1936).
114. Laun, D. D., and Corliss, C. H., *J. Res. Nat. Bur. Stand. (U.S.)*, **72A** (Phys. and Chem.) No. 6, 609–755 (Nov.–Dec. 1968).
115. Lidén, K., *Ark. Fys. (Stockholm)* **1**, No. 9, 229–267 (1949).
116. Lotz, W., *J. Opt. Soc. Am.* **57**, No. 7, 873–880 (1967).
117. Marquet, L. C., and Davis, S. P., *J. Opt. Soc. Am.* **55**, No. 5, 471–474 (1965).
118. Martin, W. C., *J. Opt. Soc. Am.* **49**, No. 11, 1071–1085 (1959).
119. Martin, W. C., *J. Res. Nat. Bur. Stand. (U.S.)*, **64A** (Phys. and Chem.) No. 1, 19–28 (Jan.–Feb. 1960) and unpublished material (1970).
120. Martin, W. C., *J. Opt. Soc. Am.* **53**, No. 9, 1047–1050 (1963).
121. Martin, W. C., and Corliss, C. H., *J. Res. Nat. Bur. Stand. (U.S.)*, **64A** (Phys. and Chem.) No. 6, 443–479 (Nov.–Dec. 1960).
122. Martin, W. C., and Kaufman, V., *J. Res. Nat. Bur. Stand. (U.S.)*, **74A** (Phys. and Chem.) No. 1, 11–22 (Mar.–Apr. 1970).
123. McConkey, J. W., and Kernahan, J. A., *Phys. Lett. (Amsterdam)* **27A**, No. 2, 82–83 (1968).

References—Continued

124. McNally, J. R., Jr., Harrison, G. R., and Park, H. B., *J. Opt. Soc. Am.* **32**, No. 6, 334–347 (1942).
125. McNally, J. R., Jr., and Vander Sluis, K. L., *J. Opt. Soc. Am.* **49**, No. 2, 200 (1959).
126. Meggers, W. F., *Rev. Mod. Phys.* **14**, Nos. 2, 3, 96–103 (1942).
127. Meggers, W. F., unpublished analysis (1959).
128. Meggers, W. F., and Moore, C. E. (Editor), *J. Res. Nat. Bur. Stand. (U.S.)*, **71A** (Phys. and Chem.) No. 6, 396–544 (Nov.–Dec. 1967).
129. Midtal, J., and Aashamar, K., *Norvegica* **2**, No. 2, 99–109 (1967).
130. Minnhagen, L., *Ark. Fys. (Stockholm)* **18**, No. 8, 97–134 (1960).
131. Minnhagen, L., *Ark. Fys. (Stockholm)* **21**, No. 26, 415–478 (1962).
132. Minnhagen, L., unpublished material (September 1967).
133. Minnhagen, L., private communication (June 1969).
134. Minnhagen, L., and Hallén, K., private communication (1969).
135. Moore, C. E., "Atomic Energy Levels," *Nat. Bur. Stand. (U.S.)*, Circ. 467, Vol. I (1949).
136. Moore, C. E., "Atomic Energy Levels," *Nat. Bur. Stand. (U.S.)*, Circ. 467, Vol. II (1952).
137. Moore, C. E., "Atomic Energy Levels," *Nat. Bur. Stand. (U.S.)*, Circ. 467, Vol. III (1958).
138. Moore, F. L., Jr., unpublished material (1967).
139. Odabasi, H., *J. Opt. Soc. Am.* **57**, No. 12, 1459–1463 (1967).
140. Ölme, A., *Ark. Fys. (Stockholm)* **40**, No. 2, 35–47 (1969).
141. Ölme, A., unpublished material (1970).
142. Ölme, A., Edlén, B., Johns, J. W. C., and Herzberg, G., See *J. Opt. Soc. Am.* **60**, No. 7, 889–891 (1970).
143. Palenius, H. P., *Ark. Fys. (Stockholm)* **37**, No. 23, 368–369 (A) (1968); **39**, No. 3, 15–69 (1968); **39**, No. 39, 425–427 (1969).
144. Palenius, H. P., First Ann. Conf. European Assoc. At. Sp. (A), Paris-Orsay (1969) and unpublished material (1969).
145. Pekeris, C. L., *Phys. Rev.* **112**, No. 5, 1649–1658 (1958); **126**, No. 4, 1470–1476 (1962).
146. Persson, W., and Minnhagen, L., *Ark. Fys. (Stockholm)* **37**, No. 22, 273–300 (1968).
147. Petersson, B., *Ark. Fys. (Stockholm)* **27**, No. 23, 317–319 (1964).
148. Poppe, R., *Physica* **40**, No. 1, 17–26 (1968).
149. Racah, G., *Bull. Research Council Israel* **5A**, 78 (1955).
150. Radziemski, L. J., Jr., Andrew, K. L., Kaufman, V., and Litzén, U., *J. Opt. Soc. Am.* **57**, No. 3, 336–340 (1967).
151. Radziemski, L. J., Jr., and Kaufman, V., *J. Opt. Soc. Am.* **59**, No. 4, 424–443 (1969).
152. Rao, Y. B., *Indian J. Phys.* **32**, No. 11, 497–515 (1958).
153. Reader, J., and Davis, S. P., Program, Atomic Sp. Symposium, *Nat. Bur. Stand. (U.S.)*, p. 79 (A) (1967).
154. Reader, J., and Davis, S. P., *J. Res. Nat. Bur. Stand. (U.S.)*, **71A** (Phys. and Chem.) No. 6, 587–599 (Nov.–Dec. 1967).
155. Reader, J., and Epstein, G. L., *J. Opt. Soc. Am.* **60**, No. 1, 140 (L) (1970).
- 155a. Reader, J., and Epstein, G. L., private communication (April 1970).
156. Reader, J., and Sugar, J., *J. Opt. Soc. Am.* **56**, No. 9, 1189–1194 (1966).
- 156a. Reader, J., and Sugar, J., *J. Opt. Soc. Am.* **60**, in press (L) (1970).
157. Reeves, E. M., Garton, W. R. S., and Bass, A., *Proc. Phys. Soc. (London)* **86**, Part 5, No. 553, 1077–1080 (1965).
158. Rico, F. R., *An. Real Soc. Esp. Fis. y Quim. (Madrid)* [A] **61**, Nos. 3, 4, 103–118 (1965).
159. Risberg, G., *Ark. Fys. (Stockholm)* **28**, No. 32, 381–395 (1965).
160. Risberg, G., *Ark. Fys. (Stockholm)* **37**, No. 18, 231–249 (1968).
161. Risberg, P., *Ark. Fys. (Stockholm)* **9**, No. 31, 483–494 (1955).
162. Risberg, P., *Ark. Fys. (Stockholm)* **10**, No. 41, 583–606 (1956).
163. Ross, C. B., Jr., Thesis, Purdue University (June 1969).
165. Russell, H. N., *J. Opt. Soc. Am.* **40**, No. 9, 550–575 (1950).
166. Russell, H. N., *J. Opt. Soc. Am.* **40**, No. 9, 618–619 (1950).
167. Russell, H. N., Albertson, W., and Davis, D. N., *Phys. Rev.* **60**, No. 9, 641–656 (1941).
168. Russell, H. N., and King, A. S., *Astroph. J.* **90**, No. 1, 155–203 (1939).
169. Schröder, J. F., Dissertation, Amsterdam (1970).
- 169a. Seaton, M. J., *Proc. Phys. Soc. (London)* **88**, Part 4, No. 562, 815–832 (1966).
170. Shenstone, A. G., *J. Opt. Soc. Am.* **44**, No. 10, 749–759 (1954).
171. Shenstone, A. G., *Canadian J. Phys.* **38**, No. 5, 677–692 (1960).
172. Shenstone, A. G., *Proc. Roy. Soc. London [A]* **261**, No. 1305, 153–174 (1961).
173. Shenstone, A. G., *J. Res. Nat. Bur. Stand. (U.S.)*, **67A** (Phys. and Chem.) No. 2, 87–112 (Mar.–Apr. 1963).
174. Shenstone, A. G., unpublished material (Dec. 1965).
175. Shenstone, A. G., and Meggers, W. F., *J. Res. Nat. Bur. Stand. (U.S.)* **61**, No. 5, 373–411 (1958) RP2908.
176. Sugar, J., *J. Res. Nat. Bur. Stand. (U.S.)*, **66A** (Phys. and Chem.) No. 4, 321–324 (July–Aug. 1962).
177. Sugar, J., *J. Opt. Soc. Am.* **53**, No. 7, 831–839 (1963).
178. Sugar, J., *J. Opt. Soc. Am.* **55**, No. 1, 33–58 (1965) and unpublished material (1970).
179. Sugar, J., *J. Opt. Soc. Am.* **55**, No. 9, 1058–1061 (1965) and unpublished material (1970).
180. Sugar, J., *J. Res. Nat. Bur. Stand. (U.S.)*, **73A** (Phys. and Chem.) No. 3, 333–381 (May–June 1969).
- 180a. Sugar, J., *J. Opt. Soc. Am.* **60**, No. 4, 454–466 (1970).
181. Sugar, J., and Reader, J., *J. Opt. Soc. Am.* **55**, No. 10, 1286–1290 (1965).
182. Svensson, L. Å., and Ekberg, J. O., *Ark. Fys. (Stockholm)* **37**, No. 7, 65–84 (1968).
183. Svensson, L. Å., and Ekberg, J. O., *Ark. Fys. (Stockholm)* **40**, No. 14, 145–163 (1969).
184. Tech, J. L., *J. Res. Nat. Bur. Stand. (U.S.)*, **67A** (Phys. and Chem.) No. 6, 505–554 (Nov.–Dec. 1963).
185. Toresson, Y. G., *Ark. Fys. (Stockholm)* **17**, No. 12, 179–192 (1960).
186. Toresson, Y. G., *Ark. Fys. (Stockholm)* **18**, No. 28, 389–416 (1960).
187. Velasco, R., and Adames, J., *Publ. Inst. Opt. (Madrid)* No. 26, 124 pp. (1966).
188. Velasco, R., and Gullón, M. N., *Opt. Pura y Aplicada (Madrid)* **1**, No. 2, 93–102 (1968).
189. Wood, D. R., and Andrew, K. L., *J. Opt. Soc. Am.* **58**, No. 6, 818–829 (1968).
190. Wyart, J.-F., Thèse, Univ. Paris (Orsay), 105 pp. (1968).
191. Yoshino, K., *J. Opt. Soc. Am.* **59**, No. 11, 1525–1526 (A) (1969).
192. Zalubas, R., *J. Opt. Soc. Am.* **58**, No. 9, 1195–1199 (1968).
193. Zalubas, R., unpublished material (1969).
194. Moore, C. E., *Nat. Bur. Stand. (U.S.)*, Spec. Publ. 306, Sections 1–4 (1969).
195. Beardon, J. A., and others, *Physics Today* **17**, No. 2, 48–49 (1964).
196. Ionov, N. I., and Mittsev, M. A., *J. Exptl. Theoret. Phys.* **13**, No. 3, 518–519 (1961); **11**, No. 4, 972–973 (L) (1960).
197. Bakulina, I. N., and Ionov, N. I., *J. Exptl. Theoret. Phys.* **9**, No. 4, 709–712 (1959).
198. Smith, D. H., and Hertel, G. R., *J. Chem. Phys.* **51**, No. 7, 3105–3107 (1969).
199. Faktor, M. M., and Hanks, R., *J. Inorg. Nucl. Chem.* **31**, No. 6, 1649–1659 (1969).
200. Taylor, B. N., Parker, W. H., and Langenberg, D. N., *Rev. Modern Physics* **41**, 375 (1969).

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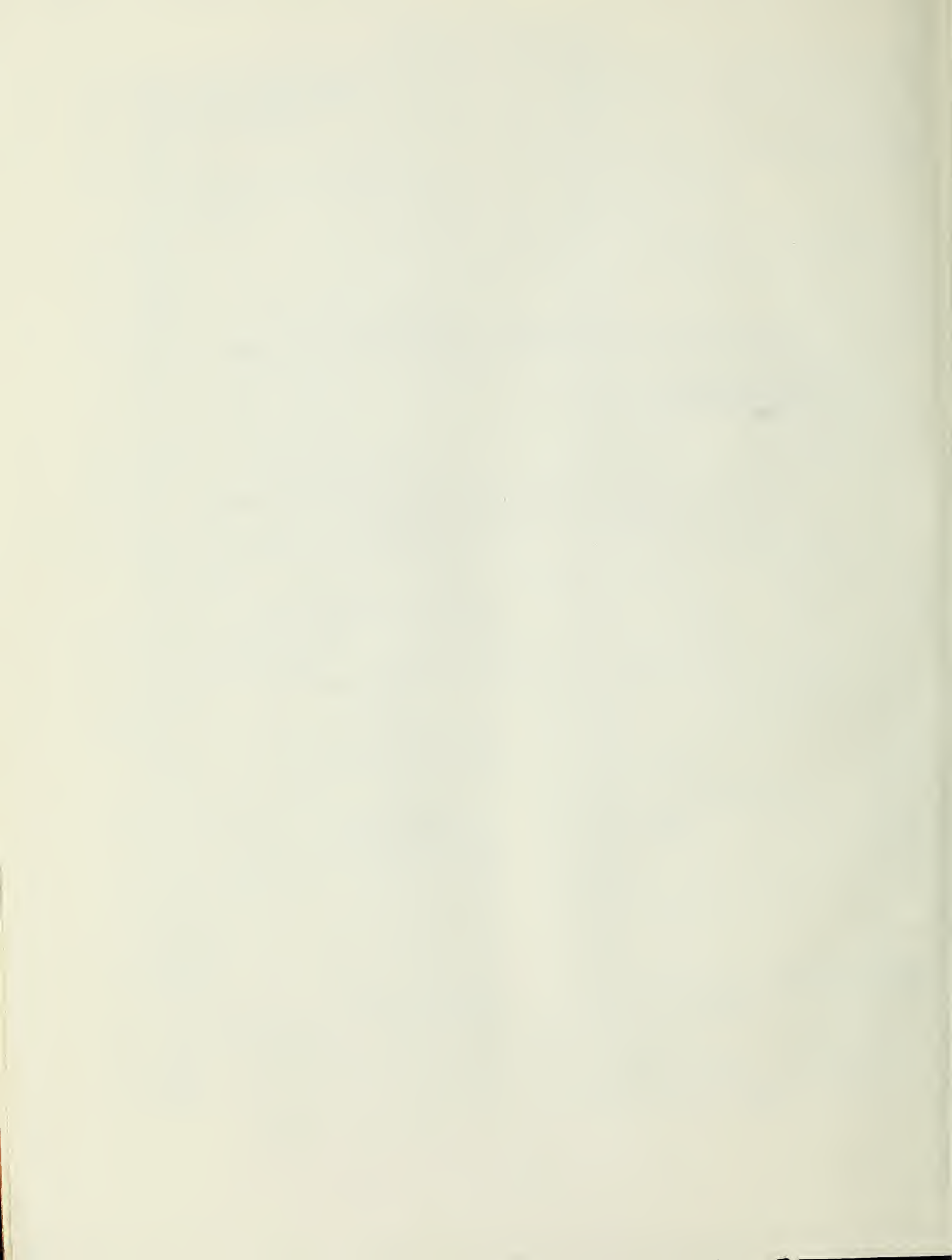
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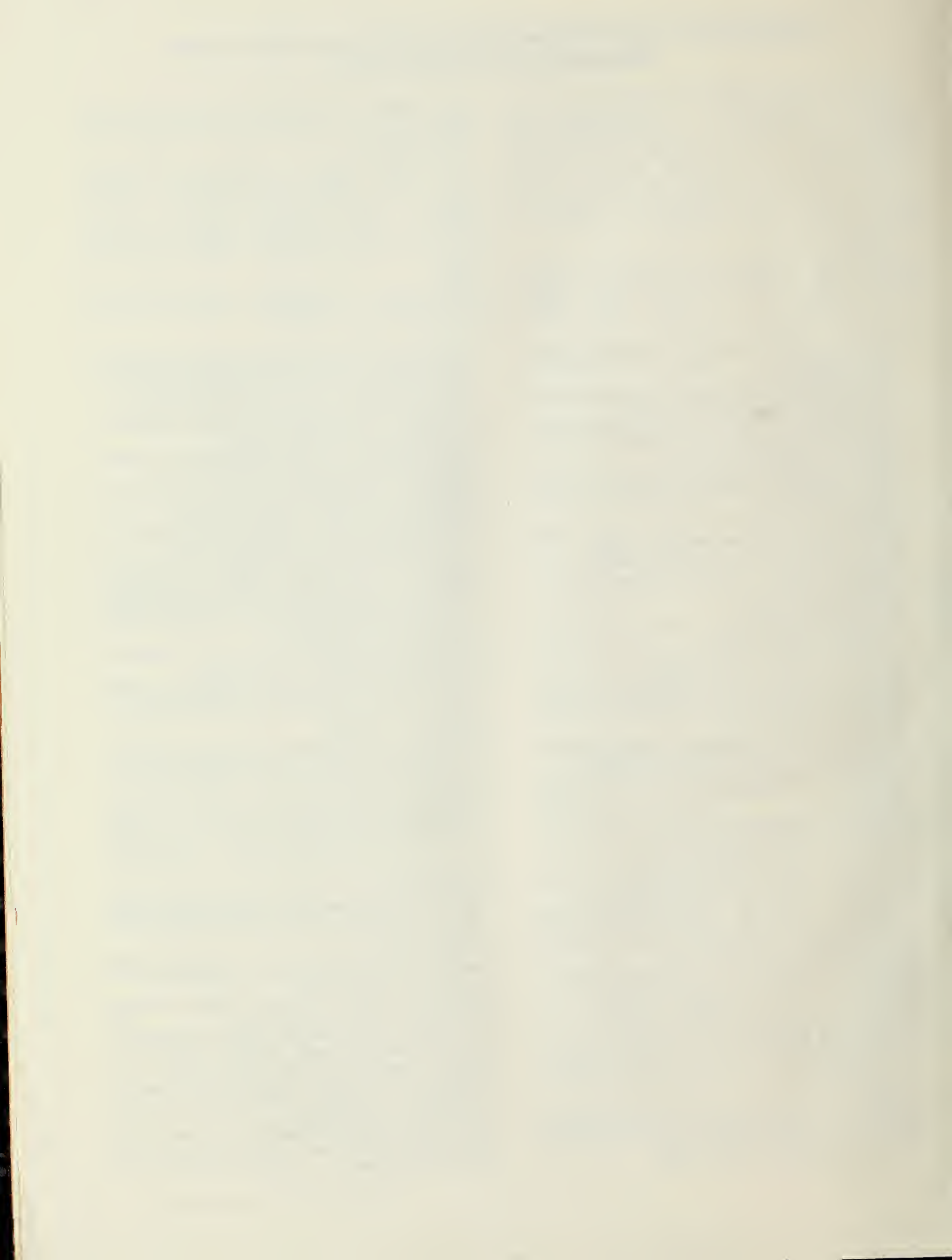
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